Aminocyclitols. XII. A Convenient Synthesis of Actinamine

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In connection with the previous paper of this series,1) a facile synthesis of actinamine has been accomplished in our laboratory.

The antibiotic actinospectacin is produced by Streptomyces spectabilis²⁾ and flavopersicus,³⁾ and inhibits Gram positive and negative bacteria. From this antibiotic, actinamine is isolated by acid hydrolysis, and its structure has been established to be N, N'-dimethyl-myo-inosadiamine-1, 3.4)

In the present study, actinamine has been prepared from 2, 4, 5, 6-tetra-O-acetyl-1, 3-di-O-ptoluenesulfonyl-myo-inositol (I)⁵⁾ in a fairly good yield.

When I was heated in boiling anhydrous hydrazine under a stream of nitrogen for 48 hr, and excess reagent was removed, a hydrazinolysis product was obtained, which was hydrogenated with Raney nickel in a hydrogen stream and

subsequently acetylated to give hexaacetyl-myoinosadiamine-1, 3 (II),65 mp 283.5-284.5°C, in 35.2% yield and hexaacetyl-muco-inosadimaine-1, 47 in 2.4% yield. A selective de-O-acetylation of II gave di-N-acetyl-myo-inosadiamine-1, 3, mp 310-311°C (dec.) in 92.0% yield. The hydrolysis of II in boiling 6 N hydrochloric acid gave myo-inosadiamine-1, 3 dihydrochloride (III) in 98.7% yield.

Then III was treated with ethyl chloroformate in an alkaline solution, and the product was acetylated to give 2, 4, 5, 6-tetra-O-acetyl-N, N'diethoxycarbonyl-myo-inosadiamine-1, 3 (IV), mp 189.5—190°C, in 85.0% yield. By reduction of IV with lithium aluminum hydride, hexaacetylactinamine (V)4,6) was obtained in 44.5% yield, after subsequent acetylation.

$$\begin{array}{c} \text{OTS} \\ \text{AcO} \\ \text{OAc} \\ \text{I} \end{array} \begin{array}{c} \text{OTS} \\ \text{OAc} \\ \text{II} \end{array} \begin{array}{c} \text{NHAC} \\ \text{OAc} \\ \text{OH} \\ \text{III} \end{array} \begin{array}{c} \text{NH2} \\ \text{OH} \\ \text{OH} \\ \text{III} \end{array} \begin{array}{c} \text{NHR} \\ \text{OAc} \\ \text{NAC} \\ \text{OAc} \\ \text{NAC} \end{array} \begin{array}{c} \text{CH3} \\ \text{NAC} \\ \text{OAc} \\ \text{NAC} \\ \text{OAc} \\ \text{NAC} \end{array} \begin{array}{c} \text{CH3} \\ \text{NAC} \\ \text{OAc} \\ \text{NAC} \\ \text{OAc} \\ \text{NAC} \end{array}$$

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